## Quest Chapter 19

| # | Problem   | Hint   |
|---|---|--|
| 1 | Which situation is likely to hurt you more?<br>1. Unable to determine   | Pressure is F/A.   |
|   | <ol> <li>Your bare foot was stepped on by a 270-lb<br/>man wearing flat-soled loafer.</li> <li>Either</li> <li>Your bare foot was stepped on by a 130-lb<br/>woman wearing high heels.</li> </ol>   | Which situation would subject you to more pressure?                                      |
| 2 | A 1.4 m wide by 2.3 m long water bed weighs   | First find the area of the bed.  |
|   | Find the pressure that the water bed exerts on<br>the floor. Assume that the entire lower surface<br>of the bed makes contact with the floor.   | Next use the equation for pressure: $P = F/A$ .  |
| 3 | At what level (vertically) should you hold a<br>cut finger to reduce bleeding?<br>1. as high as possible<br>2. at the same level as your heart<br>3. as far from your heart as possible at any<br>level   | Where will the effect of gravity reduce the bodies blood pressure the most?              |
| 4 | <ul> <li>You want a blood pressure reading as close as possible to that of your heart.</li> <li>Where should you place the cuff?</li> <li>1. anywhere the cuff will fit</li> <li>2. at a level above the heart</li> <li>3. at a level even with the heart</li> <li>4. at a level below the heart</li> </ul> | Where is the place where the person's blood pressure is the least affected by gravity?   |
| 5 | The depth of water behind the Hoover Dam<br>in Nevada is 135 m.<br>What is the water pressure at a depth of<br>135m? The weight density of water is<br>9800 N/m <sup>3</sup> .  | Weight density is<br>d <sub>w</sub> =Weight/Volume.<br>Pressure = d <sub>w</sub> * Depth |
| 6 | The pressure in a 3-foot-deep lake is P1. The<br>pressure in a 3-foot-deep hot tub 2 meters in<br>diameter is P2.<br>What relationship would P1 and P2 have?<br>1. P1 > P2<br>2. Unable to determine  | Assuming that the water is the same, will there be a difference in weight densities?     |
|   | 3. P1 = P2<br>4. P1 < P2  | equation to interpret.   |

| HINT  | <sup>!</sup>   Problem  |
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| If the whale is compressed, what happens to its volume?   | In a deep dive, a whale is appreciably<br>compressed by the pressure of the<br>surrounding water.<br>What happens to the whale's density?   |
| Does the whale's mass<br>change?  | <ol> <li>It cannot be determined.</li> <li>Its density decreases.</li> <li>Its density increases.</li> <li>Its density remains the same as before.</li> </ol>   |
| What must happen to its density if those two situations occur?  |   |
| were<br>level<br>sure<br>cm)? Normal atmospheric pressure.<br>That means the air pressure<br>on the columns of the two<br>liquids is the same and that<br>the height of the Hg column<br>will be 76cm.  | If a liquid only half as dense as mercury were<br>used in a barometer, how high would its level<br>be on a day of normal atmospheric pressure<br>(when the mercury barometer reads 76 cm)?<br>1. 19 cm<br>2. 76 cm<br>3. 38 cm<br>4. 152 cm<br>5. 304 cm  |
| Remember from #5:<br>Pressure = d <sub>w</sub> * Depth  |   |
| So, d <sub>w</sub> is halved, what happens to depth?  |   |
| <ul> <li>Does density play a part in</li> <li>to this problem? No.</li> <li>8.5</li> </ul>  | A block of aluminum with a volume of 8.5 cm <sup>3</sup> is placed in a beaker of water filled to the brim and sinks. Water overflows. The same happens in another beaker with a 8.5  |
| This is simply a matter of volume.  | cm <sup>3</sup> block of lead.<br>The lead will displace (more, less, the same<br>amount of) water than the aluminum block<br>does.   |
| How do the two volumes<br>compare?  | <ol> <li>more</li> <li>less</li> <li>the same</li> <li>It cannot be determined without a direct measurement</li> </ol>  |
| occur?were<br>level<br>sure<br>cm)?Normal atmospheric pressure<br>on the columns of the two<br>liquids is the same and that<br>the height of the Hg column<br>will be 76cm.Remember from #5:<br>Pressure = dw * DepthSo, dw is halved, what<br>happens to depth?5<br>to<br>e<br>8.5<br>ame<br>ck5<br>to<br>e<br>e8.5<br>ame<br>ck | If a liquid only half as dense as mercury were used in a barometer, how high would its level be on a day of normal atmospheric pressure (when the mercury barometer reads 76 cm)?         1. 19 cm         2. 76 cm         3. 38 cm         4. 152 cm         5. 304 cm         A block of aluminum with a volume of 8.5 cm <sup>3</sup> is placed in a beaker of water filled to the brim and sinks. Water overflows. The same happens in another beaker with a 8.5 cm <sup>3</sup> block of lead.         The lead will displace (more, less, the same amount of) water than the aluminum block does.         1. more         2. less         3. the same         4. It cannot be determined without a direct measurement. |

| #  | Problem   | Hint   |
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| 10 | Imagine holding two identical bricks under<br>water. Brick A is just beneath the surface of<br>the water, while brick B is at a greater depth.<br>What is the force needed to hold brick B in<br>place? (Assume the density of water doesn't<br>change with height.)<br>1. smaller than the force required to hold brick<br>A in place.<br>2. the same as | What determines the force<br>needed to hold the bricks in<br>place?<br>Weight and Buoyant Force.<br>What determines the size of<br>the F <sub>buoyant</sub> ? Remember |
|    | 3. larger   | Archimedes Principle.<br>So, Weight = F <sub>buoyant</sub> + F <sub>Holding</sub>  |
|    |   | Identical blocks:<br>Weight = Weight<br>Volume = Volume  |
| 11 | Suppose that a volleyball A and a bowling<br>ball B are completely submerged in water and<br>have the same volume, as in the figure. (Of<br>course, you would have to hold the volleyball   | Remember Archimedes<br>Principle.  |
|    | <ul> <li>beneath the water to keep it from popping up to the surface.)</li> <li>Which feels a greater buoyant force?</li> <li>1. They feel the same buoyant force.</li> <li>2. bowling ball B</li> <li>3. volleyball A</li> <li>4. Unable to determine</li> </ul>   | What determines the F <sub>buoyant</sub> ?   |
| 12 | Suppose that a volleyball A floats on the water,<br>and a bowling ball B, being denser than water,<br>is completely submerged in water. Assume<br>they have the same volume.  | Same balls as in #11, but the volleyball is allowed to float.  |
|    | <ul> <li>Which feels a greater buoyant force?</li> <li>1. bowling ball B</li> <li>2. volleyball A</li> <li>3. They feel the same buoyant force.</li> <li>4. Unable to determine</li> </ul>  | What has changed as you consider Archimedes principle?   |

| #   | Problem  | Hint                                   |
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| 13  | (part 1 of 2)<br>A 20-N rock hangs from a spring scale. The<br>rock is lowered into a beaker of water that sits                                | Back to Archimedes.                    |
|     | on another spring scale, but is not allowed to   | Scale A reads the effective            |
|     | touch the bottom of the beaker.  | force of gravity on the rock:          |
|     | change?  | 20N before it enters the water.        |
|     | 1. decreases correct   |  |
|     | 2. remains the same  | Do any other forces affect that        |
|     | 4. increases   | reading when it enters the             |
| 4.4 |  | water? See #10.                        |
| 14  | (part 2 of 2)<br>How does the reading on scale B change?<br>1. Unable to determine   | Remember Newton's 3 <sup>rd</sup> Law? |
|     | 2. increases   | If a buoyant force that acts on        |
|     | 3. remains the same  | the rock in the water is the           |
|     | 4. decreases   | action, what is the reaction?          |
|     |  |  |
|     |  | How would that be seen in Scale B?     |
| 15  | Why does a can of diet drink float in water,<br>while a can of regular soda sinks?<br>1. It depends on the brand and the actual<br>ingredients | Do the cans have the same volume? Yes. |
|     | 2. Sugar is heavier than a sugar substitute.   | If one floats and the other            |
|     | 3. Diet soda is less dense than a regular  | sinks and their volumes are            |
|     | 4. Diet soda cans are slightly smaller.  | the same, what must be                 |
|     | 5. Regular soda has fewer gas bubbles.   | different?                             |
|     |  | How does that difference               |
|     |  | affect the cans?                       |

| #  | Problem   | Hint  |
|----|---|---|
| 16 | Helga says that although it's impossible to<br>walk on a sea of water, it is possible to walk<br>on a sea of mercury. She claims that if you<br>step into a pool of mercury, you will only  | The human body has a density that is almost 1 g/cm <sup>3</sup> .   |
|    | <ul> <li>sink enough so that about half of your calf muscles is submerged. Ali disagrees with her statement.</li> <li>Who is right and why?</li> <li>1. Helga; mercury is much denser than water.</li> <li>2. Helga; mercury is a kind of metal but water is not.</li> <li>3. Ali; water and mercury have similar</li> </ul>  | So how would we fair in<br>mercury? (How does<br>mercury's density compare to<br>ours?)   |
| 17 | properties. You'll sink in the mercury, too.<br>When you gently push down on the pan of the<br>scale, the display show an increase in force.<br>Likewise if you do the same on the rim of a<br>beaker full of water.<br>However, what if you immerse you finger in<br>the water, without touching the beaker?<br>Then the scale reading<br>1. shows a decrease.<br>2. shows an increase.<br>3. doesn't change.  | Besides wetness, what does<br>your finger experience when<br>you immerse it in water?<br>Would that affect the reading<br>on the scale? |
| 18 | <ul> <li>When pickling cucumbers or other vegetables, it's very important to use the right amount of salt. An old recipe recommends putting an egg into the pickling solution and making sure it neither sinks nor floats: A sinking egg indicates too little salt while an egg that floats on the surface indicates too much salt.</li> <li>What is the assumption behind this recipe?</li> <li>All eggs have the same density. correct</li> <li>All eggs have the same weight.</li> <li>All eggs have the same volume.</li> <li>The salt tends to neutralize the cholesterol in the egg.</li> </ul> | What property of an egg is<br>being use to test the pickling<br>solution?   |

| #  | Problem  | Hint   |
|----|--|--|
| 19 | A piece of iron is sitting on a block of wood<br>floating in water.<br>If the iron were instead suspended beneath<br>the wood, would the wood float at the same<br>level, lower, or higher?<br>1. It depends on the ratio of iron and wood   | What would the iron<br>experience is suspended<br>below the wood that it did not<br>experience on top of it?                   |
|    | <ol> <li>Volumes.</li> <li>Higher, because there is a buoyant force acting on the iron now.</li> <li>Definitely lower, because the iron is under the wood.</li> <li>It would float only slightly lower, because the iron displaces a little water and the overall water level rises.</li> <li>It would float at the same level.</li> </ol>   | How would that affect the level at which the wood floats?  |
| 20 | Compared to an empty ship, would a ship<br>loaded with a cargo of Styrofoam sink deeper<br>or rise in the water?<br>1. It depends on the weight of the Styrofoam.<br>2. Since Styrofoam is less dense than water,<br>the ship rises.<br>3. The ship will float at the same level because<br>its density hasn't changed.<br>4. Since Styrofoam pushes down on the ship<br>with its weight, the ship sinks deeper. | What does adding all that<br>Styrofoam do to the properties<br>of the boat?<br>How would that affect the<br>level of the boat? |
| 21 | <ul> <li>A ship sailing from the ocean into a fresh water harbor sinks slightly deeper into the water.</li> <li>How does the buoyant force on it change?</li> <li>1. It doesn't change at all.</li> <li>2. It decreases slightly.</li> <li>3. It increases a lot.</li> <li>4. It decreases slightly.</li> </ul>  | What determines the buoyant for on an object?  |
| 22 | <ul> <li>How does the water level in a glass change when a floating ice cube melts?</li> <li>1. It depends on how many air bubbles were trapped inside of the ice cube.</li> <li>2. It falls.</li> <li>3. It remains unchanged.</li> <li>4. It rises.</li> <li>5. It depends on the size of the piece of ice.</li> </ul>   | How much water is displaced<br>by ice?<br>How much water is<br>"displaced" when that same<br>ice melts?                        |

| #  | Problem   | Hint  |
|----|---|---|
| 23 | <ul> <li>Why do some iron objects such as ships float<br/>when placed in water while other iron objects<br/>such as nails sink?</li> <li>1. Iron ships have large air pockets inside<br/>them, making them less dense than water and<br/>thus able to float.</li> <li>2. They are composed of different types of<br/>iron.</li> <li>3. The same reason that makes planes fly in<br/>the sky.</li> </ul> | What determines<br>"float-ability"?<br>How do the difference iron<br>differ that would allow some to<br>float and others to sink? |
| 24 | A wedge-shaped piece of wood floats in water<br>with the widest part on the bottom and the<br>narrowest part on top.<br>If we want the wood to displace the least<br>amount of water, what should we do?<br>1. Rotate it 90°.<br>2. Turn it over.<br>3. It doesn't matter.<br>4. Leave it as is.  | What determines how much<br>water is displaced by an<br>object placed in it?<br>Archimedes  |
| 25 | <ul> <li>Two identical glasses are filled to the same level with water. One of the two glasses has some ice cubes floating in it.</li> <li>Which glass weighs more?</li> <li>1. Not enough information is given.</li> <li>2. They weigh the same.</li> <li>3. The glass with ice cubes.</li> <li>4. The glass without ice cubes.</li> </ul>   | See #22.  |
| 26 | A piston A has a diameter of 0.64 cm, as<br>shown. A second piston B has a diameter of<br>3.1 cm.<br>In the absence of friction, determine the<br>magnitude of the force F necessary to support<br>the 569.4 N weight.  | Pascal's Principle.<br>Set up your ratio:<br>$F_{in}/A_{in} = F_{out}/A_{out}$<br>and solve for the unknown<br>force.             |